

# Teaching Transparency Worksheet Atomic And Ionic Radii

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T169 Teach rg Transparercy Worksh eets Answer Key Chem stry Matter ard Change

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#### **1. 2. Li: K: O: S: 3. 4. Na: Cl**

Answers to: "Atomic and Ionic Radii" - Transparency 21 1 The size of atoms increases down a group in the periodic table 2 Li:  $1s^2 2s^1$  K:

$1s^2 2s^2 2p^6 3s^2 3p^4$  S:  $1s^2 2s^2 2p^6 3s^2 3p^4$  K has a greater number of energy levels than Li, which explains the difference in size

between the 2 atoms The same is true for O and S 3

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TEACHING TRANSPARENCY WORKSHEET Isotopes 1 What do the following symbols represent? b no 2 Which subatomic particles are found in an atom's nucleus? 3 Which subatomic particle identifies an atom as that of a particular element? 4 Explain why atoms ...

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To understand matter at the atomic level, one must understand light and the interaction between light and matter Consider the diagram below It represents the beginnings of an electromagnetic (EM) spectrum below Complete the diagram by labeling the following regions: ultraviolet, infrared, x-ray, radio wave, gamma radiation, and microwave

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Name Date Class 14 Use with Chapter 10, Section 103 MATH SKILLS TRANSPARENCY WORKSHEET Calculating the Molar Mass of a Compound Determine the molar mass of each of the following compounds

**Teaching Transparency Master 2 Answer Key**

TEACHING TRANSPARENCY WORKSHEET Atomic Orbitals 1 What is the shape of an s orbital? Class 16 Use with Chapter 5, Section 52 2 What is the relationship between the size of an s orbital and the principal energy level in which it is found? 3 What is the shape of a p orbital? How many p orbitals are there in a sublevel?

**Atomic Structure and Chemical Bonds**

Atomic Structure and Chemical Bonds 7 Name Date Class Lab Preview Directions: Answer these questions before you begin the Lab 1 Where are electrons located relative to the nucleus? 2 Where are neutrons and protons located? How do they relate to an element's atomic number? As more information has become known about the structure of the atom,

**Teaching Transparency Worksheet The Periodic Table Answers**

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**TEACHING TRANSPARENCY**

TEACHING TRANSPARENCY 18 Chemistry: Matter and Change Teaching Transparency Worksheet 2 Use with Chapter 6, Section 61 The Periodic Table 1 How many elements are listed in the periodic table? \_\_\_\_ 2 What is the atomic number of selenium? \_\_\_\_ 3 What is the symbol for palladium? \_\_\_\_ 4 What is the atomic mass of strontium? \_\_\_\_ 5

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atomic mass properties periodic law atomic number Henry Moseley Dmitri Mendeleev The table below was developed by John Newlands and is based on a relationship called the law of (1) repeated every (2) element (3) According to this law, the properties of the elements elements Thus, for example, element two and have similar properties

**TEACHING TRANSPARENCY MASTER 11 Cathode Ray ...**

Teaching Transparency Worksheets Chemistry: Matter and Change • Chapter 4 91 Isotopes TEACHING TRANSPARENCY WORKSHEET Use with Chapter 4, Section 43 13 1 What do the following symbols represent? a e2 b n0 c p1 2 Which subatomic particles are found in an atom's nucleus? 3

Which subatomic particle identifies an atom as that of a

### **TEACHING TRANSPARENCY MASTER 16 Atomic Orbitals ...**

Teaching Transparency Worksheets Chemistry: Matter and Change • Chapter 5 9 1 What is the shape of an s orbital? 2 What is the relationship between the size of an s orbital and the principal energy level in which it is found? 3 What is the shape of a p orbital? How many p orbitals are there in a sublevel? 4 How many electrons can each

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atomic mass properties periodic law atomic number Henry Moseley Dmitri Mendeleev The table below was developed by John Newlands and is based on a relationship called the law of (1) repeated every (2) element (3) According to this law, the properties of the elements elements Thus, for example, element two and have similar properties